

Supporting information

Solubilities of Amino Acids in the Presence of Chaotropic Anions

Mehriban Aliyeva^{1,2,3}, Paula Brandão³, João A. P. Coutinho³, Olga Ferreira^{1,2}, Simão P. Pinho^{1,2*}

¹Centro de Investigação de Montanha (CIMO), Instituto Politécnico de Bragança, Campus de Santa Apolónia, 5300-253 Bragança, Portugal

²Laboratório para a Sustentabilidade e Tecnologia em Regiões de Montanha, Instituto Politécnico de Bragança, Campus de Santa Apolónia, 5300-253 Bragança, Portugal

³CICECO – Aveiro Institute of Materials, Department of Chemistry, University of Aveiro, 3810-193 Aveiro, Portugal

*To whom correspondence should be addressed:

Simão P. Pinho, Telephone: +351273303086, Fax: +351273313051, E-mail: spinho@ipb.pt

SI1 – Solubility data of AAs in pure water

Table S1. Summary of the literature solubility data of the amino acids in pure water, studied in this work.

Amino acid	Solubility of AAs/ g of AA/1000 g of water	Reference	Literature average \pm standard deviation
Glycine	234.350	[1]	236.7 \pm 3.6
	235.700	[2]	
	234.894	[3]	
	242.026	[4]	
	238.33	This work	
L-leucine	21.774	[5]	22.4 \pm 1.1
	24.260	[6]	
	21.600	[7]	
	21.800	[8]	
	23.243	[9]	
	21.520	[10]	
21.54	This work		
L-phenylalanine	28.000	[8]	28.4 \pm 0.9
	29.700*	[11]	
	27.800	[3]	
	27.900	[7]	
	28.35	This work	

*isomer not specified.

SI2 – Solid phase studies

The identified solid phases are compiled in Erro! Fonte de referência não encontrada. for each of the studied amino acids.

Table S2. Cell parameters for the studied amino acids from the supplier determined by single crystal X-ray diffraction and comparison with published data in CCDC Cambridge database.

Amino acid	Crystal form	CCDC code	Reference
Glycine (used with Na-tosylate)	Monoclinic P (α -glycine) a=5.106(4)Å; b=11.991(7)Å; c=5.452(3)Å; beta=111.53(4)°	1416373	Jiang <i>et al.</i> [12]
Glycine ^a (used with thiocyanates)	Monoclinic P (α -glycine) a=5.10 Å; b=11.99 Å; c=5.45 Å; β = 111.53°	1416373	Jiang <i>et al.</i> [12]
	Hexagonal P (γ -glycine) a=b=7.00 Å; c=5.48 Å	1416374	Jiang <i>et al.</i> [12]
L-Aspartic acid ^a	Monoclinic P a=5.112(7)Å, b=6.921(10)Å; c=7.602(10)Å, beta= 100.38(3)°	652520	Bendeif and Jelsch [13]
L-Phenylalanine ^a	Monoclinic P a=8.832(9)Å; b=6.064(8)Å; c=31.383(10)Å; beta= 97.36(5)°	1012155	Ihlefeldt <i>et al.</i> [14]
L-Leucine ^a	Monoclinic P A=9.631(4) Å; b=5.340(3) Å; 14.654(6) Å; beta=94.09(3)°	1508364	Binns <i>et al.</i> [15]

^aData reported in a previous work [16].

The crystal structure of the different amino acids in aqueous solutions of NaSCN, KSCN, and NH₄SCN salts or Na-tosylate were analysed by powder X-ray diffraction and compared with the simulated powder pattern obtained from the published data deposited in CCDC Cambridge database in **Fig. S2, S3, S4, and S6**.

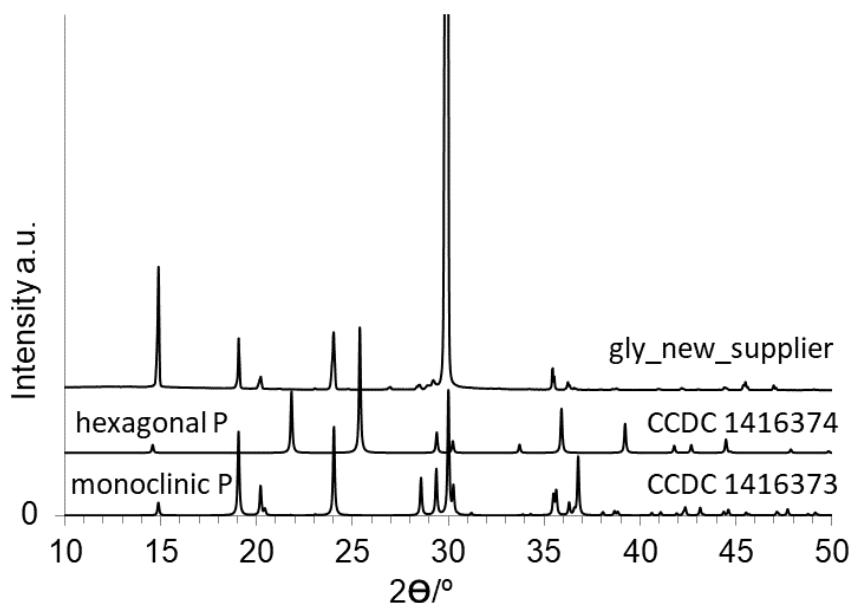


Fig. S1. Comparison of the experimental X-ray powder diffraction pattern of the solid phase samples of glycine (used in the solubility experiments with Na-tosylate) from supplier with the powder pattern calculated from the single-crystal X-ray diffraction data CCDC 1416373 (α form) and CCDC 1416374 (γ form).

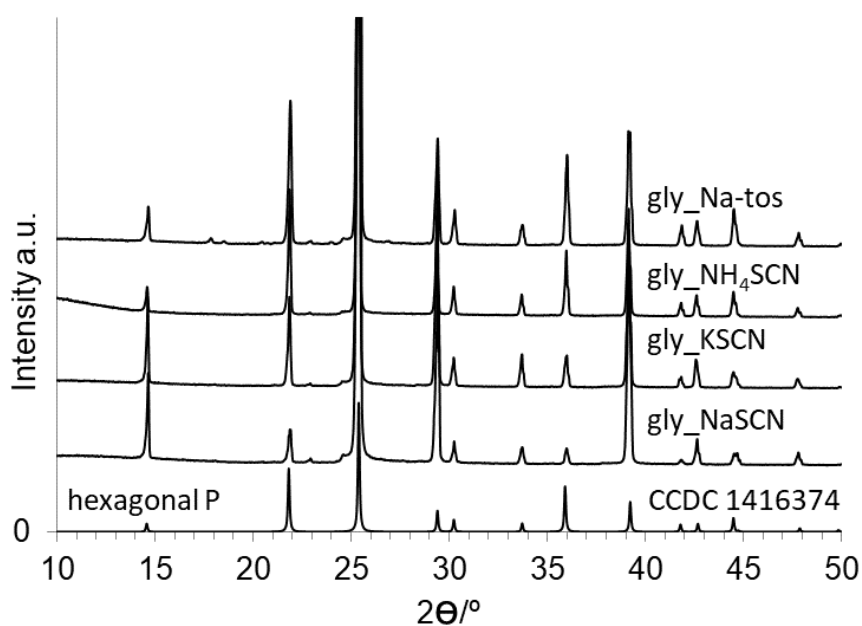


Fig. S2. Comparison of the experimental X-ray powder diffraction pattern of the solid phase samples of glycine (filtrated from aqueous solutions of NaSCN, KSCN, and NH_4SCN salts or Na-tosylate) with the powder pattern calculated from the single-crystal X-ray diffraction data 1416374 (γ form).

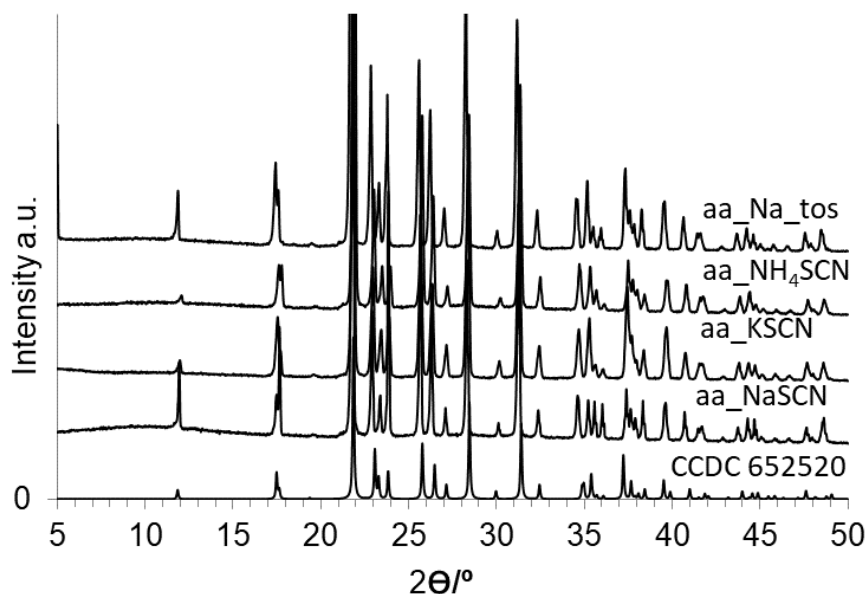


Fig. S3. Comparison of the experimental X-ray powder diffraction pattern of the solid phase samples of aspartic acid (filtrated from aqueous solutions of NaSCN, KSCN, and NH₄SCN salts or Na-tosylate) with the powder pattern calculated from the single-crystal X-ray diffraction data CCDC 652520.

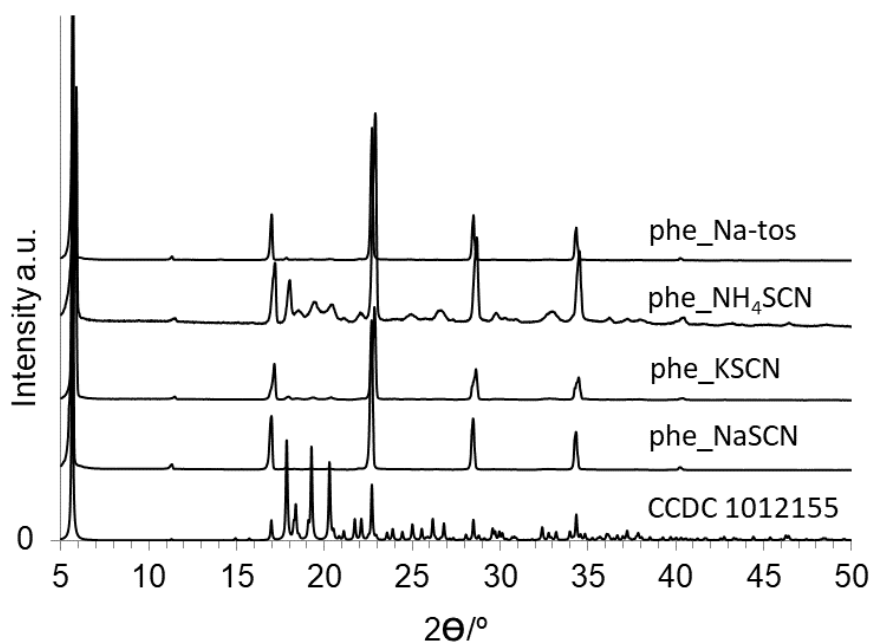


Fig. S4. Comparison of the experimental X-ray powder diffraction pattern of the solid phase samples of phenylalanine (filtrated from aqueous solutions of NaSCN, KSCN, and NH₄SCN salts or Na-tosylate at 1.5 molal) with the powder pattern calculated from the single-crystal X-ray diffraction data CCDC 1012155.

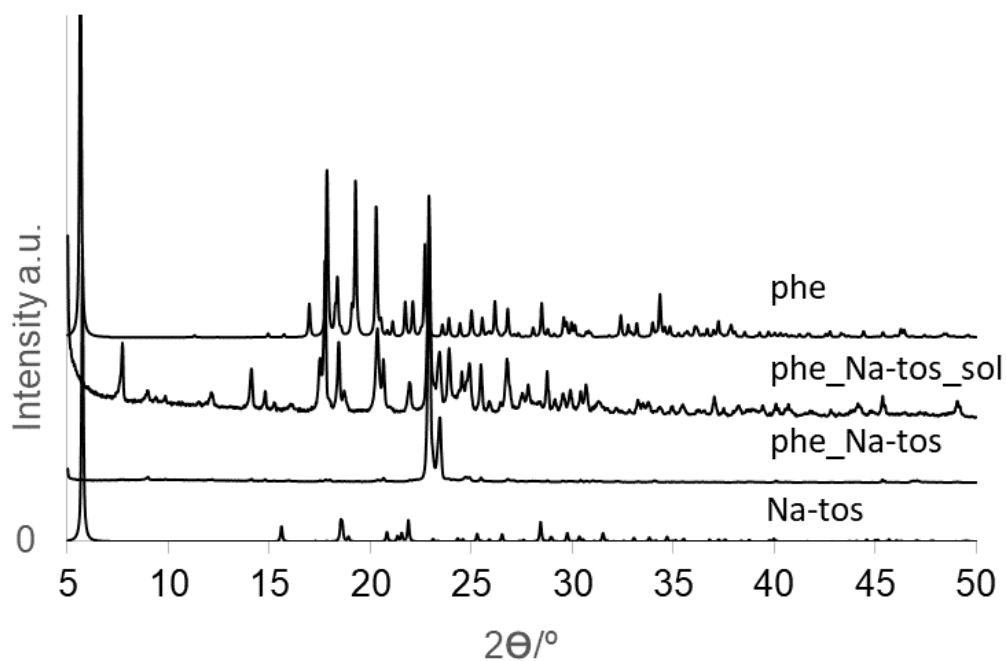


Fig. S5. Comparison of the experimental X-ray powder diffraction pattern of the solid phase samples of phenylalanine (filtrated from aqueous solutions of Na-tosylate at 2 molal) with the powder pattern calculated from the single-crystal X-ray diffraction data CCDC 1012155.

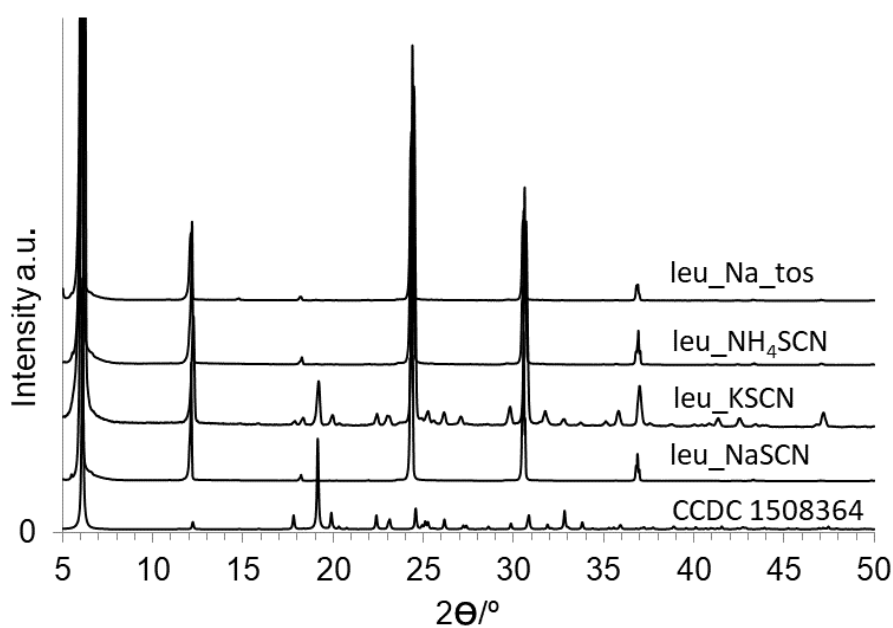


Fig. S6. Comparison of the experimental X-ray powder diffraction pattern of the solid phase samples of leucine (filtrated from aqueous solutions of NaSCN, KSCN, and NH₄SCN salts or Na-tosylate) with the powder pattern calculated from the single-crystal X-ray diffraction data CCDC 1508364.

References

1. Romero, C.M., Oviedo, C.D.: Effect of temperature on the solubility of α -amino acids and α,ω -amino acids in water. *J Solution Chem.* 42, 1355–1362 (2013). <https://doi.org/10.1007/s10953-013-0031-9>
2. Ferreira, L.A., Macedo, E.A., Pinho, S.P.: Effect of KCl and Na₂SO₄ on the solubility of glycine and DL-alanine in water at 298.15 K. *Ind Eng Chem Res.* 44, 8892–8898 (2005). <https://doi.org/10.1021/ie050613q>
3. Vasantha, T., Kavitha, T., Kumar, A., Venkatesu, P., Rama Devi, R.S.: Evaluating the transfer free energies of amino acids from water to ammonium-based ionic liquids at 298.15 K. *J Mol Liq.* 208, 130–136 (2015). <https://doi.org/10.1016/j.molliq.2015.04.007>
4. Zeng, Y., Li, Z.: Phase Equilibria for the Glycine–Methanol–NH₄Cl–H₂O System. *Ind Eng Chem Res.* 53, 16864–16872 (2014). <https://doi.org/10.1021/IE502846M>
5. Matsuo, H., Suzuki, Y., Sawamura, S.: Solubility of α -amino acids in water under high pressure: glycine, L-alanine, L-valine, L-leucine, and L-isoleucine. *Fluid Phase Equilib.* 200, 227–237 (2022). [https://doi.org/10.1016/S0378-3812\(02\)00029-8](https://doi.org/10.1016/S0378-3812(02)00029-8)
6. Dalton, J.B., Schmidt, C.L.A.: The solubilities of certain amino acids and related compounds in water, the densities of their solutions at twenty-five degrees, and the calculated heats of solution and partial molal volumes. *J Biol Chem.* 103, 549–578 (1933). [https://doi.org/10.1016/s0021-9258\(18\)75234-4](https://doi.org/10.1016/s0021-9258(18)75234-4)
7. Nozaki, Y., Tanford, C.: The solubility of aminoacids, diglycine, and triglycine in aqueous guanidine hydrochloride solutions. *J Biol Chem.* 245, 1648–1652 (1970). [https://doi.org/10.1016/S0021-9258\(19\)77141-5](https://doi.org/10.1016/S0021-9258(19)77141-5)
8. Gekko, K., Ohmae, E., Kameyama, K., Takagi, T.: Acetonitrile-protein interactions: amino acid solubility and preferential solvation. *Biochim. Biophys. Acta, Protein Struct. Mol. Enzymol.* 1387, 195 (1998). [https://doi.org/10.1016/s0167-4838\(98\)00121-6](https://doi.org/10.1016/s0167-4838(98)00121-6)
9. Carta, R., Tola, G.: Solubilities of L-cystine, L-tyrosine, L-leucine, and glycine in aqueous solutions at various pHs and NaCl concentrations. *J Chem Eng Data.* 41, 414–417 (1996). <https://doi.org/10.1021/JE9501853>
10. Thomas, D.W.: Studies on the Purification and Properties of L-leucine. Studies on the Mode of Action of Trypsin and Chymotrypsin, (1951)
11. Shi, G., Dang, Y., Pan, T., Liu, X., Liu, H., Li, S., Zhang, L., Zhao, H., Li, S., Han, J., Tai, R., Zhu, Y., Li, J., Ji, Q., Mole, R.A., Yu, D., Fang, H.: Unexpectedly enhanced solubility of aromatic amino acids and peptides in an aqueous solution of divalent transition-metal cations. *Phys Rev Lett.* 117, 1–6 (2016). <https://doi.org/10.1103/PhysRevLett.117.238102>

12. Jiang, Q., Shtukenberg, A.G., Ward, M.D., Hu, C.: Non-topotactic phase transformations in single crystals of β -glycine. *Cryst Growth Des.* 15, 2568–2573 (2015). <https://doi.org/10.1021/acs.cgd.5b00187>
13. Bendeif, E.E., Jelsch, C.: The experimental library multipolar atom model refinement of L-aspartic acid. *Acta Crystallogr C.* 63, o361–o364 (2007). <https://doi.org/10.1107/S0108270107021671>
14. Ihlefeldt, F.S., Pettersen, F.B., von Bonin, A., Zawadzka, M., Görbitz, C.H.: The Polymorphs of L-Phenylalanine. *Angewandte Chemie International Edition.* 53, 13600–13604 (2014). <https://doi.org/10.1002/anie.201406886>
15. Binns, J., Parsons, S., McIntyre, G.J.: Accurate hydrogen parameters for the amino acid L-leucine. *Acta Crystallogr B Struct Sci Cryst Eng Mater.* 72, 885–892 (2016). <https://doi.org/10.1107/S2052520616015699>
16. Aliyeva, M., Brandão, P., Gomes, J.R.B., Coutinho, J.A.P., Ferreira, O., Pinho, S.P., Ferreira, O.: Electrolyte Effects on the Amino Acid Solubility in Water: Solubilities of Glycine, L-Leucine, L-Phenylalanine, and L-Aspartic Acid in Salt Solutions of (Na⁺, K⁺, NH₄⁺)/(Cl⁻, NO₃⁻). *Ind Eng Chem Res.* 61, 5620–5631 (2022). <https://doi.org/10.1021/acs.iecr.1c04562>