

Supporting Information

Recovery of non-steroidal anti-inflammatory drugs from wastes using ionic liquid-based three-phase partitioning systems

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Table S1. General information of the pills used as real matrices in this study; identification and content of the NSAIDs , name of the medicine, list of excipients and MA holder [information available on the website of the Portuguese National Authority of Medicines and Health Products (INFARMED, I.P.) - www.infarmed.pt].

API	Medicine name (NSAIDs content)	Excipients	MA holder
NAP	<i>Naproxeno Generis</i> (naproxen 250 mg)	Magnesium stearate, microcrystalline cellulose, colloidal silica, povidone	Generis Farmacêutica, Amadora, Portugal
KET	<i>Profenid Retard</i> (ketoprofen 200 mg)	Tablet core: calcium hydrogen phosphate dihydrate, hydroxyethyl cellulose, magnesium stearate Coating: cellulose acetate phthalate, ethyl phthalate	Sanofi Aventis, Porto Salvo, Portugal
IBU	<i>Brufen</i> (ibuprofen 200 mg)	Microcrystalline cellulose, crosslinked sodium carboxymethylcellulose, lactose monohydrate, colloidal silica dioxide, sodium lauryl sulfate, magnesium stearate, hypromellose (hydroxypropyl methylcellulose) 2910, hypromellose 2910, talc and titanium dioxide	Abbott Laboratórios, Amadora, Portugal

Table S2. Validation parameters for the HPLC-DAD analytical method.

	Ibuprofen	Naproxen	Ketoprofen
Retention Time (min)	16.48 ^a	12.73	12.27
Concentration range ($\mu\text{g}\cdot\text{mL}^{-1}$)	10 – 750 ^a	10 - 750	10 - 750
Minimum of 6 concentrations	0.05 - 50	0.05 - 50	0.05 - 50
Injection volume (μL)	10 ^a	10	10
	25	25	25
	230 ^a	270	270
Wavelength (nm)	230	245	245
	0.9999 ^a	1.0000	1.0000
R²	1.0000	1.0000	1.0000
Linearity range ($\mu\text{g}\cdot\text{mL}^{-1}$)	10-750 ^a	10-750	10-750
	0.5-50.00	0.05-50.00	0.10-50.00
LOQ	10 ^a	10	10
	0.50	0.05	0.10
LOD	1 ^a	1	1
	0.20	0.02	0.05
Accuracy intra-day (%)	83.1-101.0 ^a	80.5-104.9	83.0-99.5
	92.9-101.6	98.2-112.6	89.2 - 100.5
Accuracy inter-day (%)	88.7-109.7 ^a	87.2-109.3	81.9-114.3
Precision intra-day (RSD, %)	0.15-3.00 ^a	0.04-1.73	0.02-0.50
	0.70-4.00	0.20-1.90	0.10-2.00
Precision inter-day, (%)	0.01-1.93 ^a	0.00-0.64	0.00-0.14

^a Values taken from *e Silva, F. A.; Caban, M.; Stepnowski, P.; Coutinho, J. A. P.; Ventura, S. P. M. Recovery of ibuprofen from pharmaceutical wastes using ionic liquids. Green Chem. 2016, 18, 3749-3757. DOI: 10.1039/c6gc00261g.*

Table S3. Tie-lines (TLs) and corresponding values of tie-line lengths (TLLs) information for the IL-based ABS at 25 (± 1) °C utilized in the NSAIDs extraction.

ABS	[IL] _T ^a	[salt] _T ^b	[water] _T ^c	[IL] _M ^d	[salt] _M ^e	[IL] _B ^f	[salt] _B ^g	[water] _B ^h	TLL
[N₄₄₄₄]Cl + citrate buffer (pH 7)	68.44	1.74	29.82	29.76	30.11	0.65	51.46	47.89	84.07
[BzCh]Cl + citrate buffer (pH 7)	58.34	7.72	33.94	34.88	30.05	8.95	54.73	36.32	68.18
[C₄mim]Cl + citrate buffer (pH 7)	51.05	8.13	40.82	29.93	29.93	7.82	52.75	39.43	62.13
	55.21	6.01	38.78	33.02	28.12	7.23	53.81	38.96	67.73
	55.62	5.82	38.55	35.25	26.64	5.72	56.82	37.46	71.35
[C₄mim]Cl + citrate buffer (pH 8)	52.27	7.91	39.81	29.95	30.04	7.10	52.69	40.22	63.60
[C₄mim]Cl + potassium citrate tribasic (pH \approx9)	52.95	7.17	39.88	30.10	29.94	6.25	53.72	40.03	65.94

^a Weight fraction composition (in wt%) of IL in the top phase. | ^b Weight fraction composition (in wt%) of salt in the top phase. | ^c Weight fraction composition (in wt%) of water in the top phase. | ^d Weight fraction composition (in wt%) of IL in the biphasic mixture. | ^e Weight fraction composition (in wt%) of salt in the biphasic mixture. | ^f Weight fraction composition (in wt%) of IL in the bottom phase. | ^g Weight fraction composition (in wt%) of salt in the bottom phase. | ^h Weight fraction composition (in wt%) of water in the bottom phase.

Table S4. Extraction efficiencies (EE_{NSAID} , %) and recovery toward the top phase (R_{T} , %) plus the corresponding standard deviations (σ) of the three target NSAIDs obtained using the ABS composed of 30 wt% of $[\text{C}_4\text{mim}]\text{Cl}$ + 30 wt% of potassium citrate buffer (pH 7), and with distinct NSAIDs contents (in mg NSAID *per g* ABS).

NSAID content / (mg NSAID <i>per g</i> ABS)	1	2	4
<i>Ibuprofen</i>			
$EE_{\text{NSAID}} \pm \sigma$, %	86.7 \pm 1.3	87.9 \pm 0.7	97.8 \pm 2.9
$R_{\text{T}} \pm \sigma$, %	98.2 \pm 0.1	98.8 \pm 0.0	99.2 \pm 0.3
<i>Naproxen</i>			
$EE_{\text{NSAID}} \pm \sigma$, %	90.4 \pm 2.6	98.1 \pm 2.3	92.8 \pm 6.9
$R_{\text{T}} \pm \sigma$, %	98.2 \pm 0.1	98.5 \pm 0.0	98.7 \pm 0.7
<i>Ketoprofen</i>			
$EE_{\text{NSAID}} \pm \sigma$, %	93.7 \pm 4.0	91.9 \pm 0.1	99.2 \pm 3.5
$R_{\text{T}} \pm \sigma$, %	98.7 \pm 0.1	98.9 \pm 0.1	99.1 \pm 0.5

Table S5. Extraction efficiencies (EE_{NSAID} , %) and recovery toward the top phase (R_T , %) plus the corresponding standard deviations (σ) of the three target NSAIDs obtained using distinct IL-based ABS.

IL	[C₄mim]Cl	[N₄₄₄₄]Cl	[BzCh]Cl
<i>Ibuprofen</i>			
$EE_{\text{NSAID}} \pm \sigma, \%$	97.8 \pm 2.9	88.8 \pm 8.4	90.1 \pm 6.6
$R_T \pm \sigma, \%$	99.2 \pm 0.3	100	100
<i>Naproxen</i>			
$EE_{\text{NSAID}} \pm \sigma, \%$	92.8 \pm 6.9	84.9 \pm 5.6	86.6 \pm 8.5
$R_T \pm \sigma, \%$	98.7 \pm 0.7	100	99.8 \pm 0.0
<i>Ketoprofen</i>			
$EE_{\text{NSAID}} \pm \sigma, \%$	99.2 \pm 3.5	90.2 \pm 9.6	99.6 \pm 7.9
$R_T \pm \sigma, \%$	99.1 \pm 0.5	100	99.6 \pm 0.0

Table S6. Extraction efficiencies (EE_{NSAID} , %) and recovery toward the top phase (R_{T} , %) plus the corresponding standard deviations (σ) of the three target NSAIDs obtained using the ABS composed of 30 wt% of $[\text{C}_4\text{mim}]\text{Cl}$ + 30 wt% of potassium citrate buffer at distinct temperatures (± 1 °C) and pH 7.

Temperature (°C)	15	25	35	45
<i>Ibuprofen</i>				
$EE_{\text{NSAID}} \pm \sigma$, %	90.0 \pm 2.4	97.8 \pm 2.9	91.5 \pm 2.2	91.3 \pm 0.5
$R_{\text{T}} \pm \sigma$, %	97.8 \pm 0.5	99.2 \pm 0.3	99.2 \pm 0.1	99.3 \pm 0.0
<i>Naproxen</i>				
$EE_{\text{NSAID}} \pm \sigma$, %	91.2 \pm 2.0	92.8 \pm 6.9	92.5 \pm 0.4	96.4 \pm 2.2
$R_{\text{T}} \pm \sigma$, %	97.4 \pm 0.9	98.7 \pm 0.7	99.1 \pm 0.1	99.2 \pm 0.0
<i>Ketoprofen</i>				
$EE_{\text{NSAID}} \pm \sigma$, %	94.5 \pm 1.6	99.2 \pm 3.5	94.8 \pm 1.0	100
$R_{\text{T}} \pm \sigma$, %	97.7 \pm 0.7	99.1 \pm 0.5	99.3 \pm 0.1	99.4 \pm 0.1

Table S7. Extraction efficiencies (EE_{NSAID} , %) and recovery toward the top phase (R_{T} , %) plus the corresponding standard deviations (σ) of the three target NSAIDs obtained using ABS composed of 30 wt% of $[\text{C}_4\text{mim}]\text{Cl}$ + 30 wt% of potassium citrate buffer or potassium citrate tribasic, at distinct pH values.

pH	7	8	≈9
<i>Ibuprofen</i>			
$EE_{\text{NSAID}} \pm \sigma, \%$	97.8 ± 2.9	94.2 ± 4.8	85.1 ± 3.8
$R_{\text{T}} \pm \sigma, \%$	99.2 ± 0.3	99.0 ± 0.2	99.1 ± 0.2
<i>Naproxen</i>			
$EE_{\text{NSAID}} \pm \sigma, \%$	92.8 ± 6.9	91.7 ± 0.8	86.5 ± 0.6
$R_{\text{T}} \pm \sigma, \%$	98.7 ± 0.7	98.8 ± 0.3	99.0 ± 0.2
<i>Ketoprofen</i>			
$EE_{\text{NSAID}} \pm \sigma, \%$	99.2 ± 3.5	95.3 ± 3.9	89.1 ± 0.8
$R_{\text{T}} \pm \sigma, \%$	99.1 ± 0.5	99.1 ± 0.2	99.1 ± 0.3

Table S8. Extraction efficiencies (EE_{NSAID} , %) and recovery toward the top phase (R_{T} , %) plus the corresponding standard deviations (σ) of the three target NSAIDs obtained using the ABS composed of $[\text{C}_4\text{mim}]\text{Cl}$ + potassium citrate buffer (pH 7) at variable mixture compositions (in wt%)/TLLs.

$([\text{IL}]_{\text{M}}, [\text{Salt}]_{\text{M}}) / (\text{wt}\%)$	(30, 30)	(33, 28)	(35, 26.5)
<i>TLL</i>	62.13	67.73	71.35
<i>Ibuprofen</i>			
$EE_{\text{NSAID}} \pm \sigma, \%$	97.8 \pm 2.9	85.6 \pm 3.7	80.3 \pm 2.6
$R_{\text{T}} \pm \sigma, \%$	99.2 \pm 0.3	98.2 \pm 1.4	99.4 \pm 0.2
<i>Naproxen</i>			
$EE_{\text{NSAID}} \pm \sigma, \%$	92.8 \pm 6.9	86.5 \pm 3.1	82.5 \pm 3.0
$R_{\text{T}} \pm \sigma, \%$	98.7 \pm 0.7	98.1 \pm 1.4	99.4 \pm 0.2
<i>Ketoprofen</i>			
$EE_{\text{NSAID}} \pm \sigma, \%$	99.2 \pm 3.5	93.0 \pm 1.2	85.1 \pm 2.9
$R_{\text{T}} \pm \sigma, \%$	99.1 \pm 0.5	99.0 \pm 0.8	99.4 \pm 0.2

Table S9. Comparison between the proposed stability study protocol and the original OECD 111e one.

Parameter	OECD 111e	Present protocol
Purpose	Estimating hydrolysis as abiotic degradation path in environment.	Determining ILs' influence on the stability of drugs in extraction processes.
Test substance application	Compounds with good solubility in water are required.	Compounds with good solubility in water are not mandatory. It will depend upon compounds' solubility in IL enriched matrix.
Concentration of tested compound	10^{-2} to 10^{-3} M for environmental conditions ($\leq 10^{-6}$ M).	Not confined. Similar concentrations to those instilled by the extraction process are preferable.
Temperature	50 °C	Temperature at which the IL-based ABS was conducted. Other temperatures may be included if a temperature effect study is targeted.
Time of incubation	5 days (preliminary test)	5 days
Sample pH	4, 7 and 9	pH at which the IL-based ABS was conducted. Other pH values may be included if a pH effect study is targeted.
Buffer solution	CLARK and LUBS, Citrate buffers of KOLTHOFF and VLEESCHHOUWER, Borate and phosphate mixtures of SÖRENSEN	Buffer (or salt) used as phase-forming agent in the ABS. Also samples without salt addition may be included to control salt effect on analyte stability.
Sub-samples taking time	t_0 and after 5 days.	St_0 (control) and after 5 days (St_{25} and St_{50}).

Table S10. Extraction efficiencies (EE_{NSAID} , %) and recovery toward the top phase (R_T , %) plus the corresponding standard deviations (σ) of the three target NSAIDs obtained using the IL-TPP investigated.

	[C ₄ mim]Cl	[N ₄₄₄₄]Cl	[BzCh]Cl
<i>Ibuprofen</i>			
$EE_{\text{NSAID}} \pm \sigma, \%$	92.3 \pm 5.1	89.8 \pm 5.1	86.3 \pm 5.5
$R_T \pm \sigma, \%$	97.8 \pm 0.7	100	99.8 \pm 0.1
<i>Naproxen</i>			
$EE_{\text{NSAID}} \pm \sigma, \%$	86.6 \pm 5.0	89.7 \pm 3.1	84.9 \pm 3.9
$R_T \pm \sigma, \%$	97.8 \pm 0.3	100	99.7 \pm 0.0
<i>Ketoprofen</i>			
$EE_{\text{NSAID}} \pm \sigma, \%$	99.5 \pm 6.2	83.8 \pm 7.7	86.8 \pm 4.1
$R_T \pm \sigma, \%$	98.1 \pm 0.3	100	98.3 \pm 1.0

Table S11. Isolation efficiencies (IE_{NSAID} , %) plus the corresponding standard deviations (σ) of the three target NSAIDs obtained using the distinct approaches based on the precipitation with an antisolvent.

Ratio of top phase and antisolvent volume	$IE_{\text{IBU}} \pm \sigma$, %	$IE_{\text{NAP}} \pm \sigma$, %	$IE_{\text{KET}} \pm \sigma$, %
<i>Citric acid aqueous solution at 25 wt%</i>			
1:4	78.7 ± 2.2	79.1 ± 3.0	-
<i>Aluminium sulphate aqueous solution at 15 wt%</i>			
1:4	-	-	76.2 ± 1.8
1:6	-	-	80.9 ± 0.7
1:8	-	-	83.4 ± 4.0
1:10	-	-	87.1 ± 0.2
1:12	-	-	87.9 ± 0.3

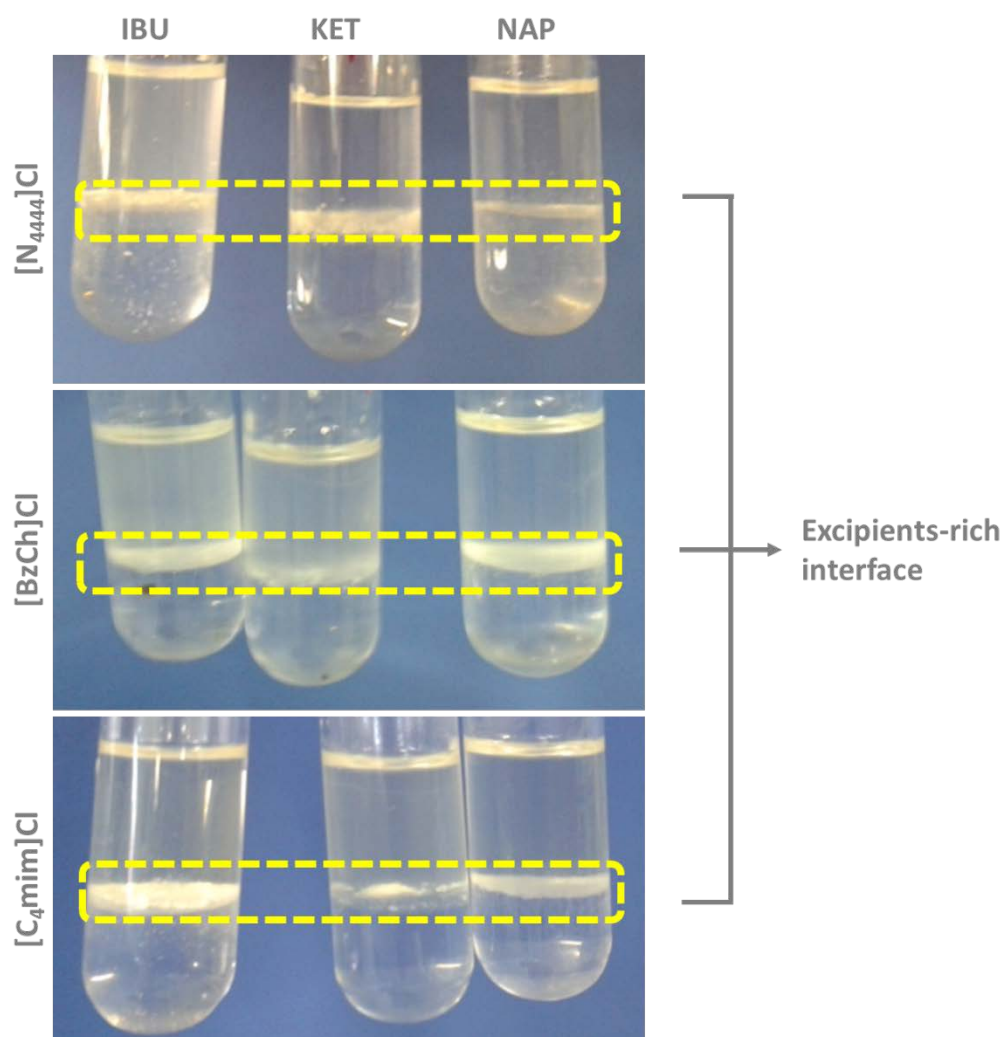


Figure S1. Visual appearance of IL-based TPP developed in the present study for the recovery of active ingredients, namely ibuprofen (IBU), ketoprofen (KET) and naproxen (NAP) from solid state pills.