

Evaluating CPA equation of state predictive capacities: A study on the transferability of the hydroxyl group associative parameters

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Supplementary Data

Table A1– Deviations for the calculated pure compound Cp.

Compound	T range (K)	%AAD Cp
MEG set 1	330-490	4.95
MEG set 2	330-490	4.92
MEG Yeh et al.	330-490	1.34
1,3-propanediol	293-353	2.20
1,4-butanediol	293-353	7.77
1,5-pentanediol	293-353	4.63
1,5-pentanediol ^a	358-586	1.15
1,6-hexanediol	319-353	2.67
Glycerol	330-530	15.31
2-propanol	293-433	1.69
2-butanol	254-354	1.14
2-pentanol	251-361	3.99
tert-butanol	254-424	4.82

a- 1,5-pentanediol results when compared with the correlation from Rowley et al.

Table A2– Deviations for the VLE at constant pressure.

Components	%AAD			
	Bubble temperatures	y ₁	Dew temperatures	x ₁
MEG + ethanol	0.16	1.38	0.68	0.43
MEG + 2-propanol	0.41	0.59	0.81	18.15
MEG + 1,3-propanediol	0.18	-	-	-
MEG + 1,4-butanediol	0.31	4.35	0.37	6.77
2-propanol + 1,3-propanediol	0.48	0.72	0.08	10.60
ethanol + tert-butanol	0.01	0.36	0.01	0.34
glycerol + methanol	0.30	-	-	-
glycerol + ethanol	2.05	-	-	-
glycerol + 1-propanol	1.47	-	-	-
glycerol + 2-propanol	1.31	-	-	-
glycerol + 1-butanol	0.50	-	-	-

Table A3– Deviations for the VLE at constant temperature/composition.

Components	%AAD			
	Bubble pressures	y ₁	Dew pressures	x ₁
tert-butanol + butane	4.44	-	1.47	-
tert-butanol + isobutylene	2.17	3.20	1.67	3.82
glycerol + 1,3-propanediol	19.52	-	-	-

Table A4– Comparison between the binary results for MEG + alkanes using the estimated k_{ij} and the correlated k_{ij} .

Component (HC)	%AAD			
	Solubility of MEG in HC		Solubility of HC in MEG	
	original k_{ij}	correlated k_{ij}	original k_{ij}	correlated k_{ij}
methane	46.62	45.35	3.65	5.34
ethane	-	-	3.64	7.86
propane	-	-	2.57	5.94
hexane	22.35	15.95	21.72	41.45
heptane	7.07	16.02	5.40	38.30
nonane	34.45	42.78	6.75	21.93

Table A5– Deviations for the remaining LLE and GLE.

Component 2	%AAD	
	MEG rich phase	Component 2 rich phase
MCH	8.33	15.29
CO ₂	9.57	27.23
N ₂	3.76	-
COS	6.91	-
H ₂ S	12.66	-